

REACTIONS OF THE N,N-DIALKYL-PYRIDYL-CARBOXYLIC AMIDES WITH LITHIUM AMIDES.
REGIOSELECTIVE LITHIATION OF N,N-DIISOPROPYL-PYRIDYL-CARBOXYLIC AMIDES, A USEFUL
METHOD FOR SYNTHESIS OF 2,3- AND 3,4-DISUBSTITUTED PYRIDINES¹

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The behaviour of the N,N-dimethyl-, N,N-diethyl- and N,N-diisopropyl-pyridylcarboxylic amides in the reactions with Et_2NLi and iPr_2NLi , and a convenient way of synthesis of 2,3- and 3,4-disubstituted pyridines, by the direct lithiation of N,N-diisopropylamides leading subsequently to the ortho new C-C bonds formation, are described.

The direct metallation reaction of pyridines has not been extensively studied, except in a few special cases². Recently Meyers observed ortho lithiation of 4-(4,4-dimethyloxazoliny-2-yl)pyridine by MeLi ³ and 3-(4,4-dimethyloxazoliny-2-yl)pyridine by 2,2,6,6-tetramethylpiperidine lithium amide⁴, which formed 3-lithio and 4-lithio derivatives, respectively. With other organolithium reagents 1,2- or 1,4-additions to the pyridine nucleus were observed^{3,4,5}. Ferles⁶ reported that ethyl esters of nicotinic and isonicotinic acids were lithiated by iPr_2NLi at 4- and 2-positions, respectively.

In this communication we wish to describe our preliminary study of the behaviour of N,N-dialkylpyridylcarboxylic amides⁷ 1a, 1b, 1c, 2a, 2b, 2c, 3a, 3b and 3c in the reactions with Et_2NLi and iPr_2NLi .

N,N-Dimethylamides 1a, 2a and 3a upon reaction with Et_2NLi at room temperature in Et_2O (1 hr.) were transferred into N,N-diethylamides⁸ 1b, 2b and 3b with yields 75%, 35% and 70%, respectively. No other products were detected. The observed transamidation process probably occurs via intermediate A. Adducts similar to A have been described in reactions of N,N-dialkylamides with lithiohydrocarbons^{2,10}. However, N,N-dimethylnicotinamide 2a in addition to the product

of the exchange reaction $\underline{2a} \rightarrow \underline{2b}$, afforded the regioselectively ortho lithiated amide $\underline{2d}$ (14%). Quenching with D_2O or reaction with Ph_2CO converted $\underline{2d}$ into amide $\underline{2a}$ or lactone^{8,9} $\underline{5}$ (11%). Longer reaction time (5 hr.) increased the yield of transformation $\underline{2a} \rightarrow \underline{2b}$ to 60%, but did not improve the yield of the lithiation reaction, only formation of ketone^{8,9} $\underline{8a}$ (6%) was observed. N,N-Diethylamide $\underline{2b}$ lithiated by Et_2NLi at room temperature in Et_2O (1 hr.) gave $\underline{2f}$, which was converted to $\underline{2g}$ (11%) or $\underline{5}$ (9%).

In reaction $\underline{2a}$ with Et_2NLi , formation of $\underline{2b}$, $\underline{2d}$ or $\underline{8a}$, the fact that $\underline{2f}$ was not observed, and the low yield for $\underline{2a} \rightarrow \underline{2d}$ and $\underline{2b} \rightarrow \underline{2f}$ suggest that reaction with lithium amide led to unreactive intermediate A. This is in agreement with the observation that increasing molar ratio of reactants from 1:1 to 1:2 did not affect the lithiation process.

N,N-Diethylamides $\underline{1b}$, $\underline{2b}$ and $\underline{3b}$ reacted with more sterically hindered iPr_2NLi at -45° in Et_2O (1 hr.) gave in good yields the regioselectively lithiated, highly reactive, derivatives $\underline{1d}$, $\underline{2f}$ and $\underline{3d}$, which were instantly transferred into the ketones^{8,9} $\underline{7}$, $\underline{8b}$ and $\underline{9}$.

Good yields in the lithiation reaction and absence of N,N-diisopropylamides $\underline{1c}$, $\underline{2c}$ and $\underline{3c}$, which would have been formed in the exchange process, indicate that in this case the intermediate type A was not formed probably for steric reasons.

In contrast with the N,N-diethylamides $\underline{1b}$, $\underline{2b}$ and $\underline{3b}$, the N,N-diisopropylamides $\underline{1c}$, $\underline{2c}$ and $\underline{3c}$ display synthetic utility since the carbonyl groups are sufficiently protected from the above described self reaction.

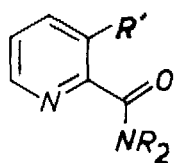
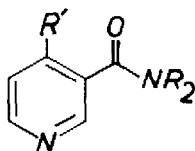
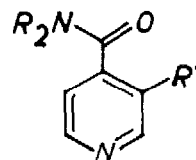
N,N-Diisopropylamides $\underline{1c}$, $\underline{2c}$ and $\underline{3c}$ lithiated by iPr_2NLi in Et_2O at -78° (2 hr.) afforded the lithiated amides $\underline{1e}$, $\underline{2h}$ and $\underline{3e}$, which were formed regioselectively and in good yields. The lithiated amides $\underline{1e}$, $\underline{2h}$ and $\underline{3e}$ reacted with Ph_2CO gave corresponding carbinols^{8,9} $\underline{1g}$, $\underline{2j}$ and $\underline{3g}$, which after acidic hydrolysis at room temperature were converted into the lactones^{8,9} $\underline{4}$, $\underline{5}$ and $\underline{6}$.

These reactive lithio derivatives could be needed in the realization of synthetic scheme for the preparation of 2,3- and 3,4-disubstituted pyridines bearing functional groups with the C-C bonds easily transformable into other derivatives.

Thus, the lithio-amides $\underline{1e}$, $\underline{2h}$ and $\underline{3e}$ in the reaction with DMF gave the formyl derivatives⁹ $\underline{1h}$, $\underline{2k}$ and $\underline{3h}$. Obtained aldehydes were reacted, without isolation, with malonic acid in the Doebner modification of the Knoevenagel reaction to give easily isolable unsaturated acids^{8,9} $\underline{1i}$, $\underline{2l}$ and $\underline{3i}$.

The recent observation by Beak¹¹, that the N,N-diethylamide function is a very good activating group for ortho-metallation of aromatics, is in agreement with results described herein.

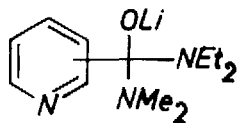
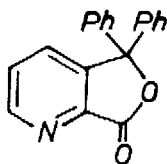
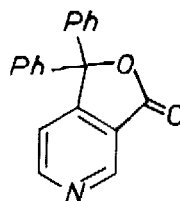
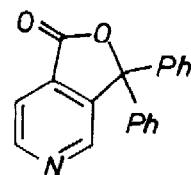
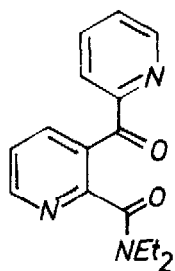
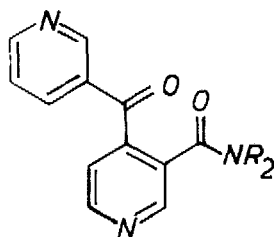
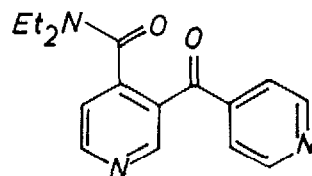
We wish to emphasize that, the N,N-dialkylamide function has a special utility in the pyridine series, since, for the compounds examined by us,

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	R	R ¹
<u>a</u> ,	Me	H
<u>b</u> ,	Et	H
<u>c</u> ,	iPr	H
<u>d</u> ,	Et	Li
<u>e</u> ,	iPr	Li
<u>f</u> ,	iPr	D (55%)
<u>g</u> ,	iPr	CPh ₂ OH (81%)
<u>h</u> ,	iPr	CHO (35%)
<u>i</u> ,	iPr	CH=CH-COOH (29%)

	R	R ¹
<u>a</u> ,	Me	H
<u>b</u> ,	Et	H
<u>c</u> ,	iPr	H
<u>d</u> ,	Me	Li
<u>e</u> ,	Me	D
<u>f</u> ,	Et	Li
<u>g</u> ,	Et	D
<u>h</u> ,	iPr	Li
<u>i</u> ,	iPr	D (46%)
<u>j</u> ,	iPr	CPh ₂ OH (68%)
<u>k</u> ,	iPr	CHO (39%)
<u>l</u> ,	iPr	CH=CH-COOH (33%)

	R	R ¹
<u>a</u> ,	Me	H
<u>b</u> ,	Et	H
<u>c</u> ,	iPr	H
<u>d</u> ,	Et	Li
<u>e</u> ,	iPr	Li
<u>f</u> ,	iPr	D (43%)
<u>g</u> ,	iPr	CPh ₂ OH (55%)
<u>h</u> ,	iPr	CHO (37%)
<u>i</u> ,	iPr	CH=CH-COOH (31%)

A4 (77%)^{8,9}5 (63%)^{8,9}6 (51%)^{8,9}7 (94%)^{8,9}89 (72%)^{8,9}

a, R = Me
b, R = Et (68%)^{8,9}

no addition to the azomethine bond was observed, even in the reaction with lithiohydrocarbons¹².

The scope and limitation of the method will be given in a full paper.

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References and Notes

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7. N,N-Dialkylamides 1a, 1b, 1c, 2a, 2c, 3a, 3b and 3c were prepared in the following sequence of reactions: Acids were chlorinated in the reaction with SOCl_2 in benzene and after removal of excess chlorinating agent, were reacted, without isolation, with the appropriate amine to afford corresponding amides. The yields of obtained products for pure materials were higher than 65%.
8. Satisfactory analytical and spectral data were obtained for all compounds.
9. Yields of obtained compounds were calculated in proportion to the starting amides; in the cases of aldehydes 1h, 2k and 3h from NMR data, for other products from isolated pure materials.
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12. N,N-Dimethylpyridylcarboxylic amides reacted with lithiohydrocarbons such as MeLi, nBuLi, iPrLi and PhLi gave with good yields respective ketones, and no addition to the pyridine nucleus was observed; J.Epsztajn, Z.Berski, J.Z.Brzeziński and A.Jóźwiak, in preparation.

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